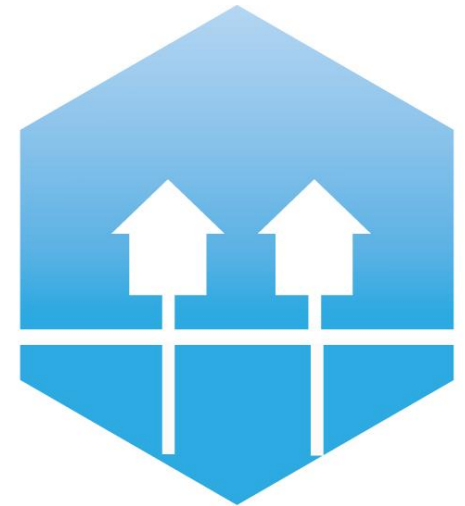


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Exergy Meters in District Heating Systems



Oliver Martin-Du Pan



AALBORG UNIVERSITY
DENMARK

4DH

4th Generation District Heating
Technologies and Systems

What is Exergy



- Exergy is defined as the theoretical maximum work which can be obtained from an energy system.
 - High grade energy (work or electricity)
 - Low grade energy (Heat)
- A simple example:
 - 1 MWh of electricity can give 1 MWh of heat.
 - 1 MWh of heat always gives less than 1 MWh of electricity because of the notion of irreversibility and the concept of entropy.



Exergy Background

- Nicolas Leonard Sadi **Carnot** in 1824 proposed a theoretical thermodynamic cycle providing an upper limit on the efficiency that any engine can achieve during the conversion of heat into work.
- Josiah Williard **Gibbs** (1839 – 1903) was accredited with the availability of energy concept by indicating that the environment plays an important part.
- Zoran **Randt** (1904 – 1972) proposed in 1956 the exergy term that quantifies in a coherent way the quantity and the quality of different forms of energy.

Carnot Factor

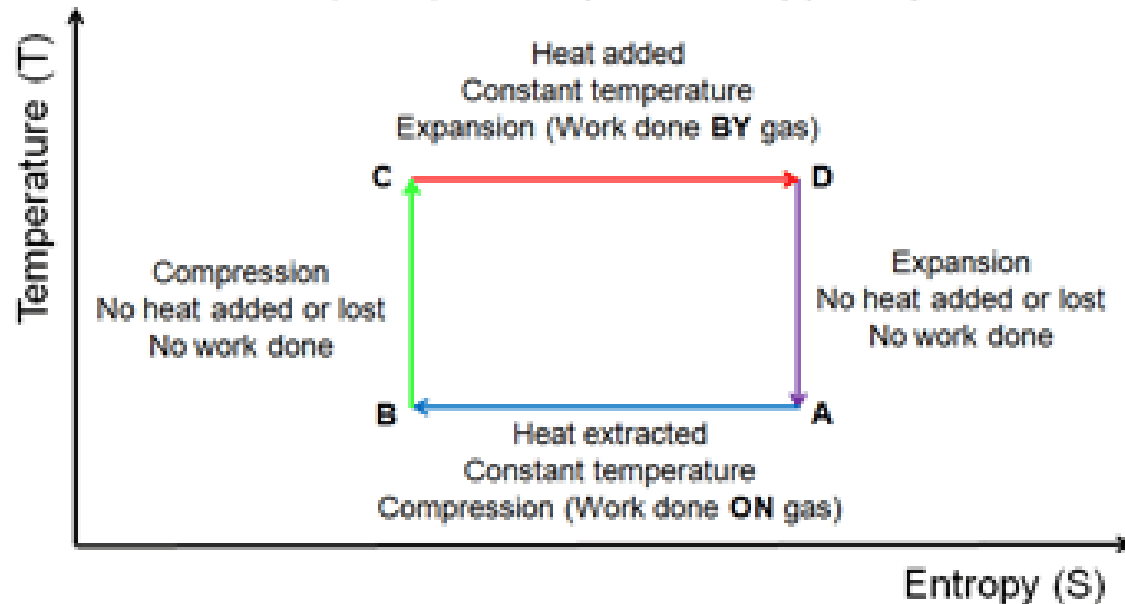
The maximum work that can be supplied by a reversible machine only depends on the temperature of the internal energy considered.

The Carnot factor or the maximum efficiency possible for any engine is:

$$\Theta = 1 - \frac{T_c}{T_h}$$



Carnot (Ideal) Heat Cycle Entropy Diagram



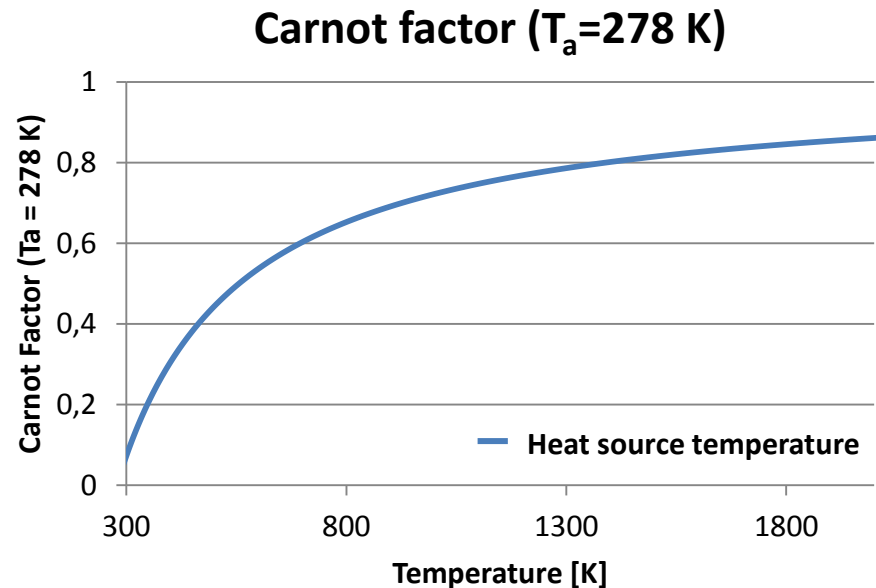
Heat Exergy

Heat exergy is the maximum work that can be obtained from a heat source by means of a reversible cycle operating between the temperatures T_i of the heat source and T_a of the atmosphere.

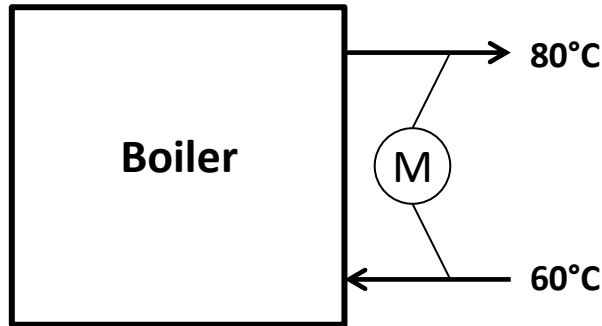
$$\dot{E}_q = \Theta * \dot{Q}_i = \left(1 - \frac{T_a}{T_i}\right) * \dot{Q}_i$$

Θ : Carnot factor

\dot{Q}_i : Heat



Exergy Meters



The exergy meter would monitor the transformation exergy \dot{E}_y cumulated on the flow leaving a heat exchanger.

$$\dot{E}_y = \dot{Q}^+ * \left(1 - \frac{T_a}{\hat{T}_{ln fluid}}\right) \quad \hat{T}_{ln fluid} = \frac{T_{out} - T_{in}}{\ln\left(\frac{T_{out}}{T_{in}}\right)}$$

If 10 MWh of cumulated heat is metered from the (M) heat meter, 1.9 MWh of Exergy would instead have been monitored by an exergy meter with an outside temperature of 5°C.

$$\dot{E}_y = 10 * \left(1 - 278 * \frac{\ln\left(\frac{353}{333}\right)}{353 - 333}\right) = 1.9 MWh$$



Heat Pump Performance



Heating coefficient of performance (COP) – (Heating effectiveness)

$$\varepsilon_h = \frac{\textit{Heat generated}}{\textit{Electricity consumption}} > 1$$

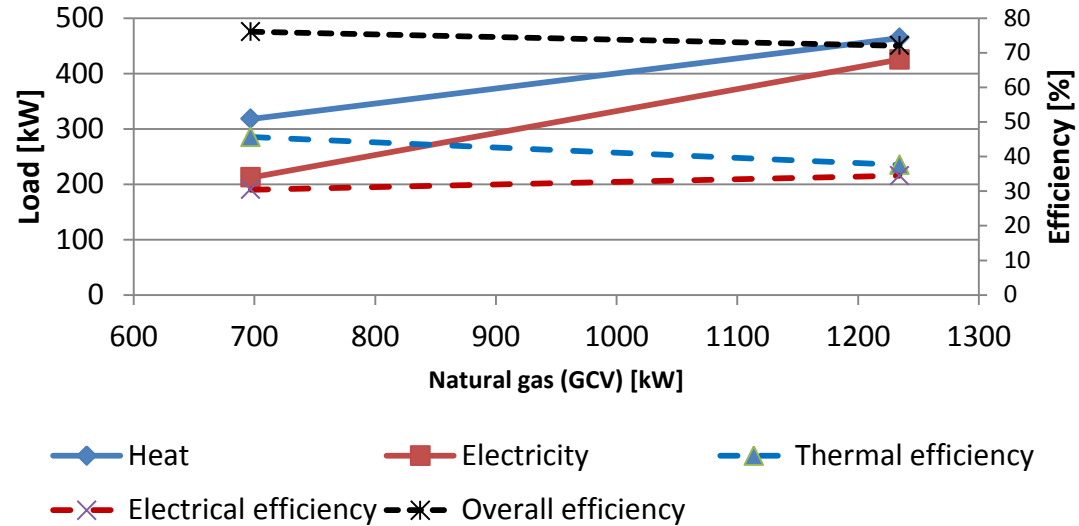
Exergy efficiency (Includes all consumptions and generation)

$$\eta = \frac{\textit{Heat exergy generated}}{\textit{Electricity consumption (exergy)}} < 1$$

Please note that the heat exergy consumed to evaporate the heat pump refrigerant equals zero.



CHP engines Performance (E-425)



Overall energy efficiency (HHV)

$$\text{Part load: } \varepsilon = \underbrace{45.6\%}_{\text{Heat exergy}} + \underbrace{30.4\%}_{\text{Electricity exergy}} = 76.1\%$$

$$\text{Full load: } \varepsilon = \underbrace{37.6\%}_{\text{Heat exergy}} + \underbrace{34.4\%}_{\text{Electricity exergy}} = 72.0\%$$

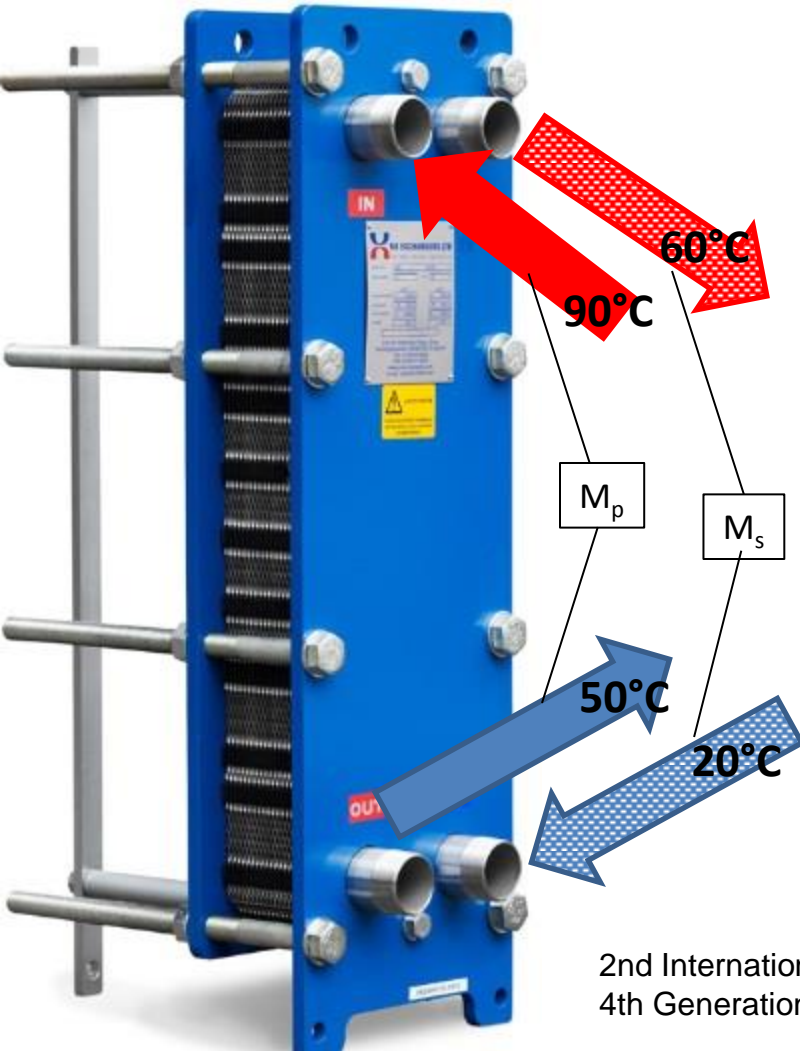
Exergy efficiency ($T_a = 5^\circ\text{C}$)

$$\text{Part load: } \eta = \underbrace{10.8\%}_{\text{Heat exergy}} + \underbrace{32.3\%}_{\text{Electricity exergy}} = 43.2\%$$

$$\text{Full load: } \eta = \underbrace{8.9\%}_{\text{Heat exergy}} + \underbrace{36.6\%}_{\text{Electricity exergy}} = 45.5\%$$

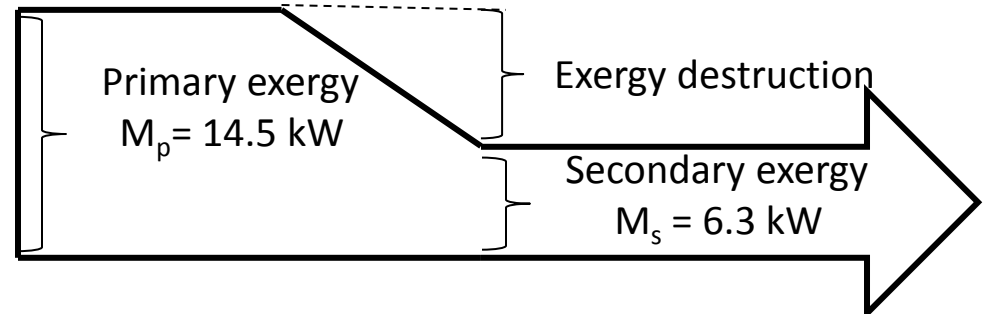
- 1) The overall energy efficiency reduces with an increasing load.
- 2) The exergy efficiency increases with an increasing load.

Plate heat exchanger and exergy destruction @ $T_a = 20^\circ\text{C}$



- 100 kW of heat from the primary flow gives 100 kW of heat on the secondary side.
- M_p : 100 kW of heat equals to 14.5 kW of exergy-heat on the primary side.
- M_s : 100 kW of heat equals to 6.3 kW of exergy-heat on the secondary side.

$$\dot{E}_y = \dot{Q}^+ * \left(1 - \frac{T_a}{\hat{T}_{\ln fluid}}\right) \quad \hat{T}_{\ln fluid} = \frac{T_{out} - T_{in}}{\ln\left(\frac{T_{out}}{T_{in}}\right)}$$



Useable Heat and Heat Meter Location

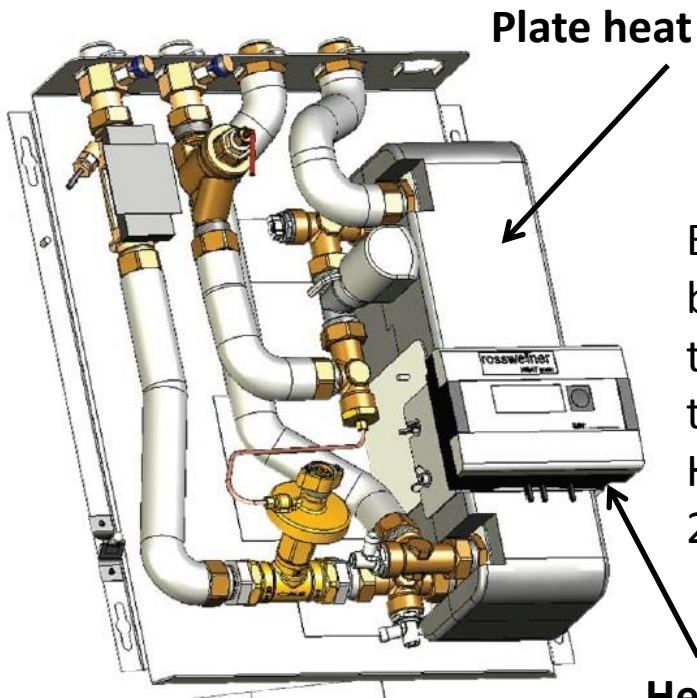


Plate heat exchanger

ESCo sell useable heat which can vary in quality. To heat a building at 21°C, the DH flow temperature must be above that temperature. However, a space would heat up sooner if the heat is supplied at a higher temperature than 21°C. Hence, heat at 70°C could be more expensive than heat at 21°C.

Heat meter : It calculates the heat consumed on the primary side. Thus, it measures the flow rate and the supply and return temperatures of the primary side.



End User

Underfloor heating

$$T_{\text{sup}} = 45^{\circ}\text{C}$$
$$T_{\text{ret}} = 30^{\circ}\text{C}$$



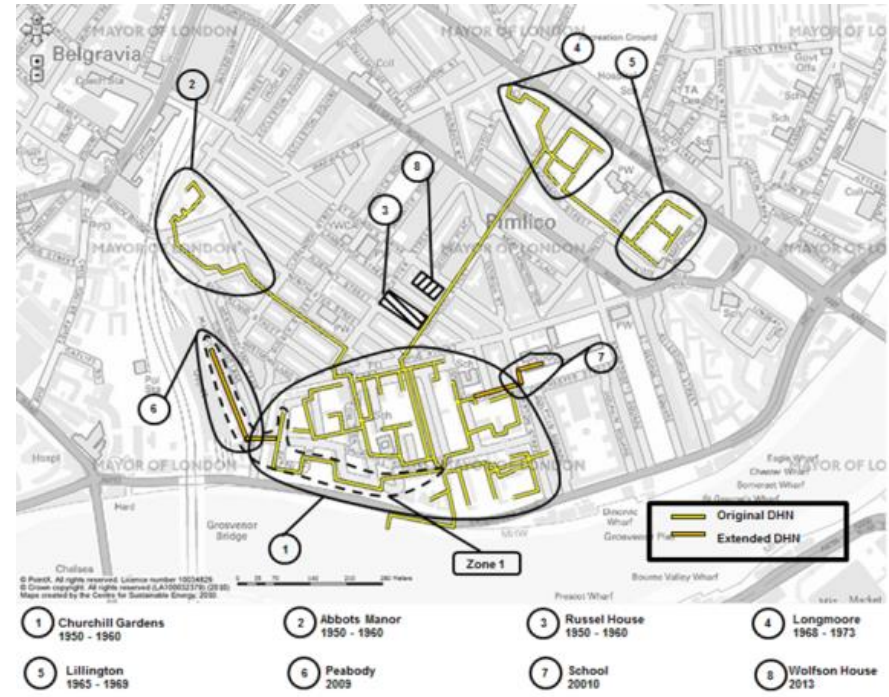
Radiator

$$T_{\text{sup}} = 60^{\circ}\text{C}$$
$$T_{\text{ret}} = 45^{\circ}\text{C}$$



Pimlico District Heating Undertaking

50,000 MWh of heat per year

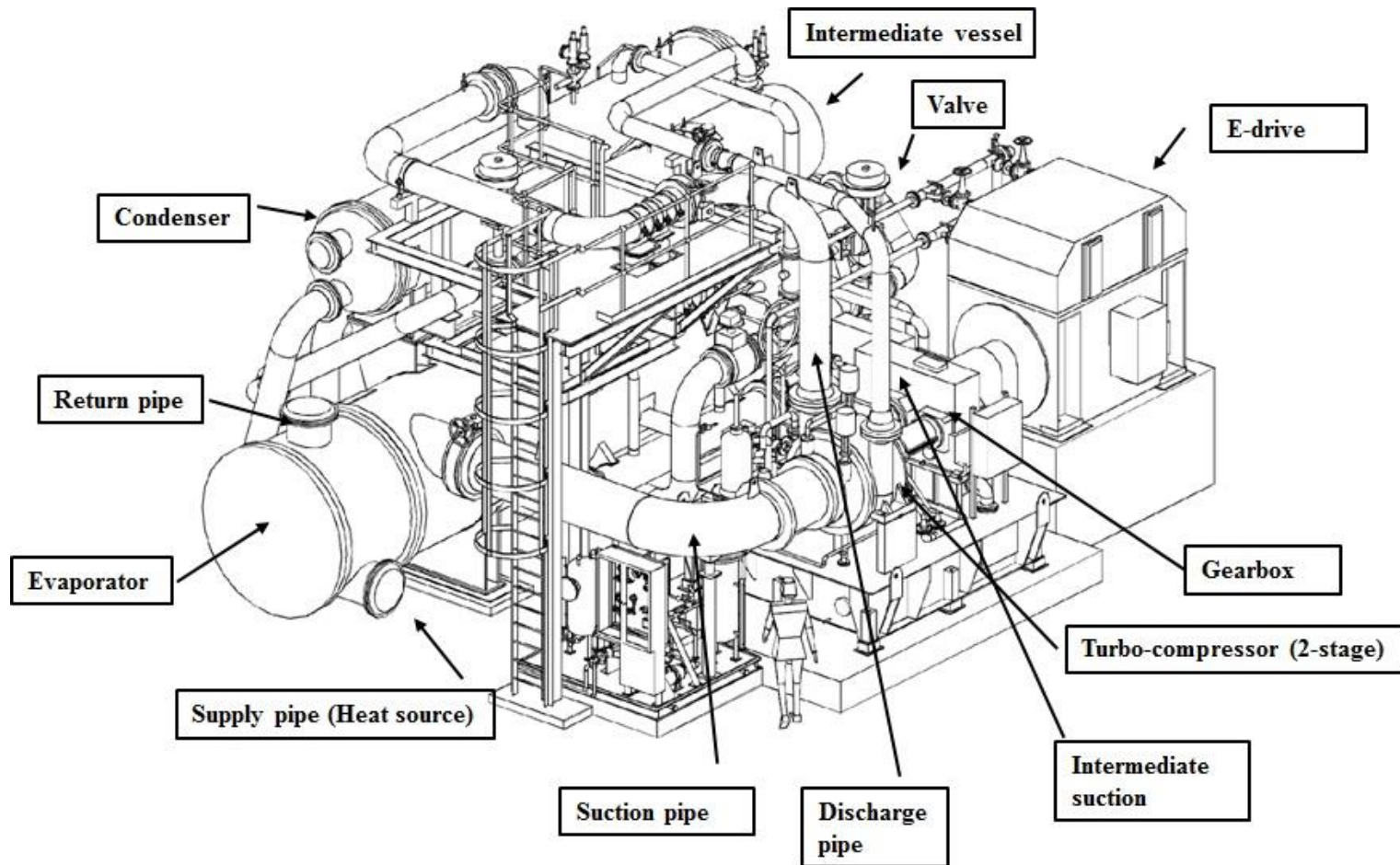


Open Loop Heat Pump

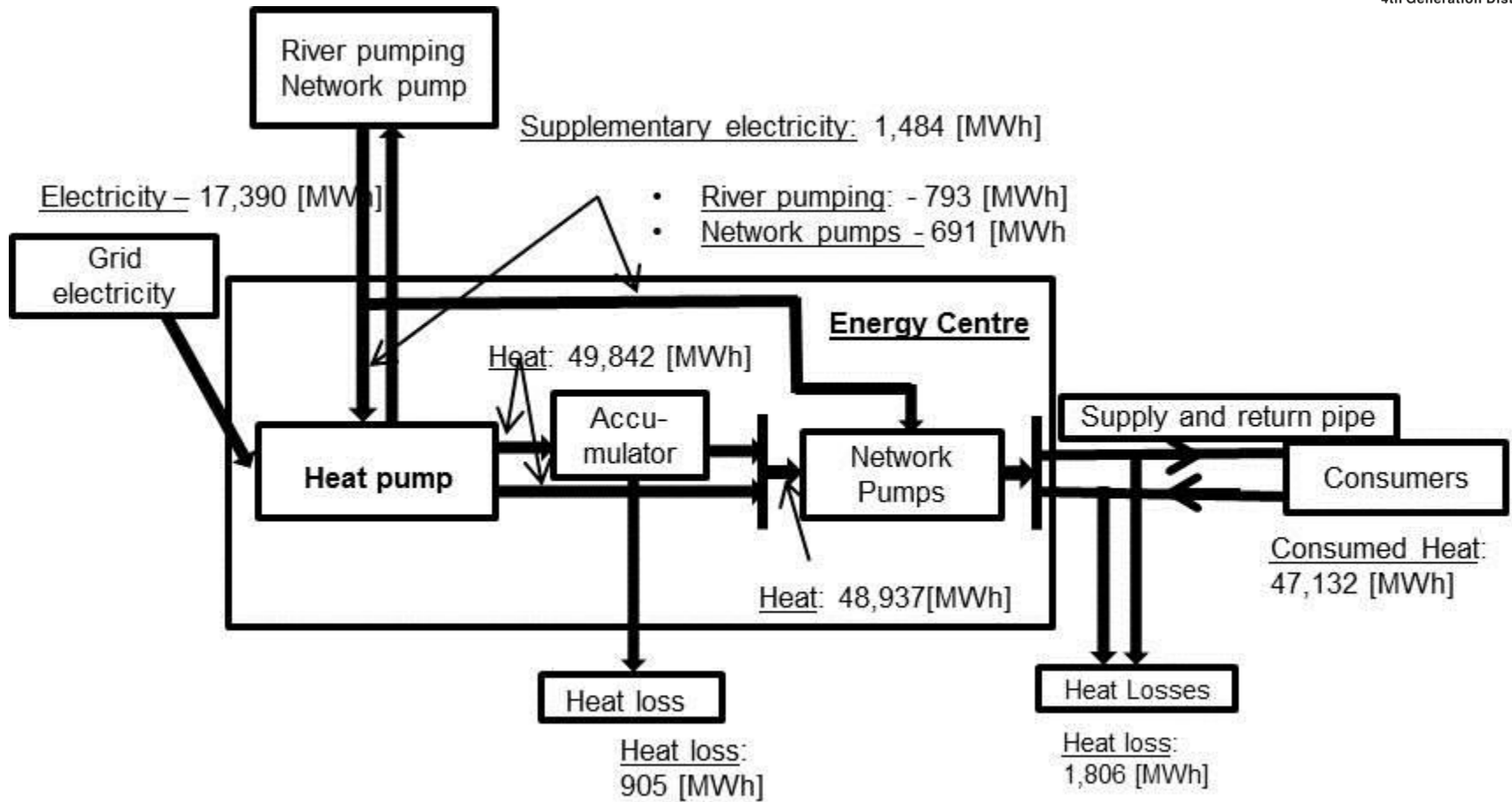


4DH

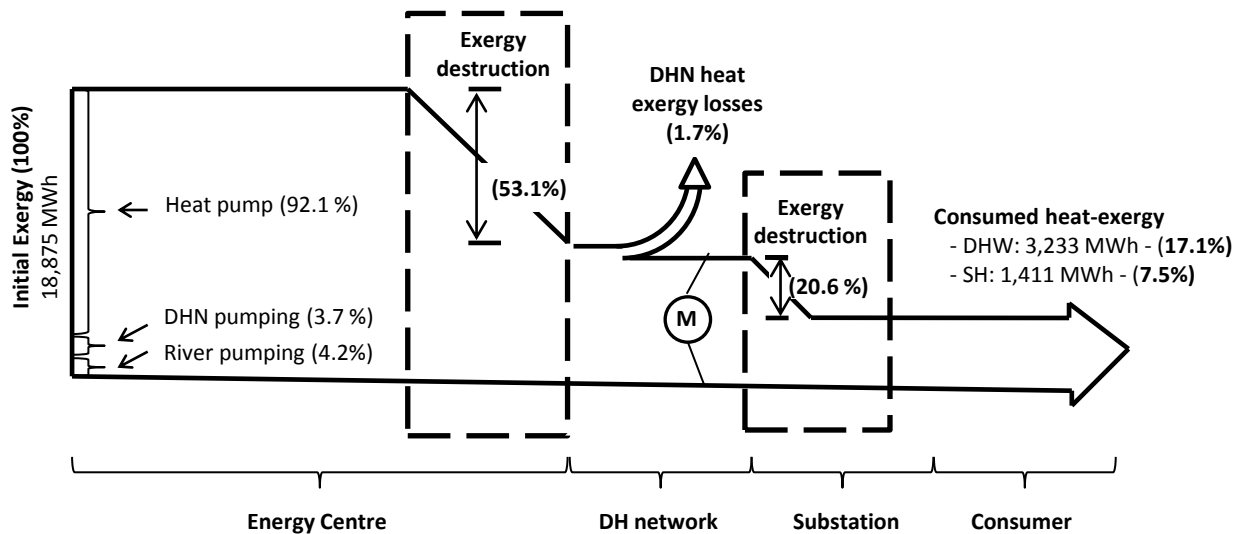
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Sankey Diagram



Grassman Diagram



In Conclusion

Exergy meters would encourage consumers to reduce their heating cost by requiring a lower flow temperature and cooling it further. This would simultaneously improve the performance of a district heating system by:

- Reducing the electricity consumption for Pumping;
- Flattening the heating load;
- Reducing the heat losses in the DH network.

Questions



Oliver Martin-Du Pan
Performance Engineer
Operations and Maintenance
T: +44 (0)78 133 69 246
M : +44 (0)2476 193836
oliver.martin-du-pan@eon-uk.com

E.ON Energy Solutions Ltd
Community Energy
47-53 Charterhouse Street
London
EC1M 6PB

